Inhibition of *Streptococcus suis* Adhesion by Dendritic Galabiose Compounds at Low Nanomolar Concentration

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A series of mono-, di-, and tetravalent galabiose (Gal α 1-4Gal) compounds were synthesized in good yields by coupling of a general carboxylic acid-bearing sugar building block to dendritic scaffolds based on the 3,5-di-(2-aminoethoxy)benzoic acid branching unit. Furthermore, a poly-(amidoamine)- (PAMAM-) based dendritic galabioside was synthesized containing eight galabiose units. All galabiosides were tested in a hemagglutination assay and a surface plasmon resonance (SPR) competition assay in order to establish their potency in the binding to the bacterial Gram-positive pathogen *Streptococcus suis*. A monovalent galabioside containing a short spacer was used as a reference compound in all the assays. Variations in the scaffold as well as in the spacer arms were introduced to determine their influence on the inhibition. The best inhibitor of hemagglutination was an octavalent galabioside with a minimal inhibitory concentration (MIC) of 0.3 nM, to the best of our knowledge the first example of inhibition of bacterial binding by a soluble carbohydrate at a subnanomolar concentration.

Introduction

Bacterial resistance toward traditional antibiotics is still increasing at an alarming rate.¹ For this reason the development of new and effective preventive methods of bacterial infections is a major priority. The first step toward an infection is the adherence of the bacterial pathogens to the epithelial surfaces of host tissues. Many of these bacterial pathogens adhere to the host via protein-carbohydrate interactions.² Inhibitors of the binding of bacterial lectin-like receptors (adhesins) to oligosaccharides present on the cell surface are potential novel antibacterial agents.^{3,4} It is likely that, because the bacteria are not killed, bacterial resistance toward these inhibitors will evolve more slowly. Moreover, resistant mutants are likely to be less infective since they have also lost their ability to bind to the tissue cells.

The adhesin–carbohydrate interaction is normally rather weak (*milli*molar concentration),³ but this is amplified by the fact that both binding partners are present in multivalent arrays.⁵ Several groups have shown that, with synthetic multivalent inhibitors, these weak interactions can also be amplified, which has already resulted in high-affinity inhibitors, especially for AB₅ toxins.⁶ In the area of bacterial anti-adhesion, relatively few studies using multivalent inhibitor design have been reported to date. Most of them used type I fimbriated *E. coli* as their target.⁷ In a recent study, good inhibition was observed. It was found that a lipophilic attachment to mannose ligands already led to a 135-fold enhancement of the inhibitory power over α -D-mannopyranoside. However, multivalency by at-

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taching these spacered mannose moieties to proteins or dendrimers only led to additional enhancements of up to 1 order of magnitude, when compared on a per sugar basis.8 Autar et al.9 previously reported studies involving multivalent GalNAc*β*1-4Gal-containing compounds targeting the pathogens F1C fimbriated E. coli and Pseudomonas aeruginosa. As in the case of E. coli, spacer effects (3-fold) combined with a modest multivalency effect (6.3-fold) led to a maximum overall enhancement of 19-fold per sugar for a divalent system. Furthermore, with Streptococcus suis, Hansen et al.¹⁰ observed a relative increase in inhibitory potency of 50fold per carbohydrate by using small di-, tri-, and tetravalent galabiose-containing molecules in hemagglutination inhibition experiments. The Gram-positive bacterium S. suis can cause meningitis, septicemia, and pneumonia in pigs and also meningitis in humans.^{11,12} S. suis recognizes the galabiose epitope (Gal α 1-4Gal) on the terminal and internal positions of various glycolipids, wherein the disaccharide acts as the ligand for pathogen adhesion to cells, as the first step of an infective process. Other pathogenic microbes that bind to the galabiose epitope are the uropathogenic E. coli via its PapG adhesin¹³ and the bacterium Pseudomonas aeruginosa via its PA-IL lectin.¹⁴ Furthermore, glycolipids bearing the galabiose epitope can bind to bacterial enterotoxins, for example, the Shiga-like toxins (SLTs)¹⁵ and verotoxin from E. coli.¹⁶ For S. suis, extensive specificity studies have been performed which showed that galabiose is the best disaccharide ligand.¹⁷ There are two galabiose-binding subtypes of S. suis designated as P_N and P_O , based on the nature of their adhesin specificity.¹⁸ Galabiose was shown to form hydrogen bonds to both adhesins via its HO-2, HO-3, HO-4', and

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Scheme 1. Synthesis of Galactose Acceptor 7^a



^{*a*} (a) 3-Bromopropanol, 0 °C, BF₃ Et₂O, 84%; (b) NaN₃, DMF, 100 °C, 87%; (c) NaOMe, MeOH, CH₂Cl₂, 97%; (d) α,α -dimethoxy-toluene, *p*-TsOH, DMF, 50 °C, 80%; (e) NaH, TBAI, BnBr, DMF, 75%; (f) TES, TFA, CH₂Cl₂, 0 °C, 76%.

HO-6', while for the $P_{\rm O}$ subtype additional weak hydrogen-bonding interactions were detected to HO-6 and HO-3'. 18,19

Though the specificities of the carbohydrate binding proteins on bacterial surface are characterized in detail, their relative geometry, in contrast to the structurally well-characterized AB₅ toxins,²⁰ is mostly unknown. To succeed in inhibitor design, it is likely important to make a structurally diverse set of inhibitors. We here report the synthesis of a diverse set of multivalent galabiose derivatives and their evaluation as inhibitors of S. suis. The length and chemical composition of the spacer arms, as well as the nature of the scaffold molecules themselves, were aspects of structural variation. The spacer arms for the compounds described here were considerably longer than those previously reported, since the longer spacers were shown to be more effective.¹⁰ Furthermore studies on AB₅ toxin inhibition reported longer spacers to be generally advantageous.²⁰ The compounds were evaluated by a hemagglutination assay and a surface plasmon resonance competition assay.

Results and Discussion

Synthesis. All multivalent galabiose derivatives and monovalent reference compounds were synthesized by

Scheme 2. Synthesis of Galabiose Building block 13^a

attachment of the carboxylic acid-containing galabioside **13** (Scheme 2). Compound **13** was synthesized by coupling of 3-azidopropyl 2,3,6-tri-O-benzyl- β -D-galactopyranoside acceptor **7** (Scheme 1) with 2,3,4,6-tetra-O-benzyl- α -D-galactopyranoside trichloroacetimidate donor **8** (Scheme 2).^{21,22} In our strategy, benzyl protective groups were used to "arm" the O4 position of the galactose acceptor and to favor the α -glycosylated disaccharide. Key building block **13** was coupled to the amine-bearing scaffolds by peptide coupling methods to give the target compounds after sugar deprotection.

The first target in the synthesis was the benzylprotected acceptor 7, which contains a free HO-4 for the glycosidic coupling and an anomeric propyl azide group for further elaboration. As a first step in the synthesis of 7, 3-bromopropanol was coupled to the anomeric center of 2,3,4,6-tetra-O-acetyl- β -D-galactose 1, in the presence of boron trifluoride ethyl etherate $(BF_3 \cdot Et_2O)$ as a catalyst, to give 2 in 84% yield. An azido function was introduced via $S_N 2$ displacement of the bromine, using sodium azide in N,N-dimethylformamide (DMF). Deacetylation of **3** by use of sodium methoxide in MeOH, followed by benzylidenation with α, α -dimethoxytoluene and *p*-toluenesulfonic acid in DMF, gave **5** in 80% yield. Benzylation of the two remaining free hydroxyl functions of 5 by use of benzylbromide, sodium hydride, and tetrabutylammonium iodide (TBAI) in DMF yielded 6 in 75% yield. Selective opening of the benzylidene ring to the O6 position by use of triethylsilane (TES) and trifluoroacetic acid (TFA)²³ gave the desired acceptor 7 in 76% yield. This reaction was also performed with sodium cyanoborohydride and HCl, which gave slightly lower yields. Coupling of the known "armed" imidate donor $\mathbf{8}^{21,22}$ to acceptor 7 in Et₂O and 15% trimethylsilyl trifluoromethanesulfonate (TMSOTf) as a catalyst gave disaccharide 9 in 73% yield (Scheme 2). Due to the solvent effect of $Et_2O^{21,24}$ and the low temperature, no β -linked disaccharide was isolated. To facilitate deprotection at the final multivalent stage, the benzyl protecting groups were replaced by acetyl groups. Selective reduction of the azido function, by Pd-CaCO₃ (Lindlar catalyst) and H_2 , followed by the introduction of the N-tert-butyloxycarbonyl (Boc) protecting group on the free amino group, gave 10 in 92% yield.²⁵ Debenzylation of 10 with 10% Pd-C and H_2 and subsequent acetyla-



^{*a*} (a) 15% TMSOTf, Et₂O, 0 °C, 73%; (b) Lindlar catalyst, H₂, Boc₂O, NEt₃, MeOH, EtOAc, 92%; (c) Pd-C, HOAc, H₂, MeOH, EtOAc; (d) pyridine, Ac₂O, 88% over two steps; (e) TFA, H₂O, CH₂Cl₂, 0 °C, quant; (f) diglycolic anhydride, pyridine, dioxane, 90 °C, quant.





 a (a) $Pd^{(0)}(PPh_3)_4,$ CuI, Et_3N, CH_3CN, 76% (16) or 87% (18); (b) TFA, CH_2Cl_2, quant (18) or quant (19).

tion by a 1:1 mixture of pyridine and acetic anhydride gave **11** (88% yield over two steps). Removal of the Bocprotecting group of **11** with trifluoroacetic acid gave compound **12**, which was converted to the key building block **13** by use of diglycolic anhydride in pyridine and dioxane at 90 °C (quantitative over two steps).⁹

To increase the hydrophobicity and rigidity of the spacer arms, in addition to the 3,5-di-(2-aminoethoxy)benzoic acid-based compounds, two scaffolds 17 and 19 were prepared that included an acetylene moiety (Scheme 3). To fix distinct geometries, 1,3- and 1,4-disubstituted benzene rings were used as scaffold elements. Bocprotected propargylamine was coupled to two commercially available scaffolds, 1,4-diiodobenzene 14 and 1,3-diiodobenzene 15, in good yields by a Sonogashira coupling with Pd⁰(PPh₃)₄, CuI, and Et₃N in acetonitrile. During the purification of 16, the monosubstituted compound was also isolated as a minor side product (16% yield). Removal of the Boc-protecting groups of compounds 16 and 18, with TFA in CH₂Cl₂, yielded acetylene-containing scaffolds 17 and 19, both in guantitative vield.

Galabiose derivative 13 was attached to mono-, di-, and tetravalent dendritic scaffolds based on the 3,5-di-(2-aminoethoxy)benzoic acid branching unit.^{26,27} These dendrimers have been used as scaffolds for the synthesis of multivalent carbohydrate compounds.^{9,28} A general coupling and deprotection route was used to produce mono-, di-, and tetravalent galabiosides 22, 25, 32, and 35 (Scheme 4). Compound 13 was coupled in good yields to the respective (poly)amino compounds by use of BOP [(Benzotriazolo-1-yloxy)tris(dimethylamino)phosphonium hexafluorophosphate] as a coupling reagent and DIPEA (*N*,*N*-diisopropylethylamine) as a base. Finally, deacetylation with sodium methoxide in MeOH and purification on a C8-cartridge column gave target compounds 22 (80%), 25 (99%), 32 (77%), and 35 (79%). Characterization by 1D and 2D NMR spectroscopy and high-resolution mass spectrometry (HRMS) confirmed the identity of the expected products. By the same protocol, the acetylenic scaffolds 17 and 19 were converted to their corresponding divalent galabiosides 27 and 29. According to their ¹³C NMR spectra, the acetylene carbon signals were still present (82 and 86 ppm), indicating that no cyclization to heterocyclic structures took place.^{29,30} Poly(amidoamine) (PAMAM) dendrimer scaffolds³¹ have extensively been used in the synthesis of dendritic carbohydrate compounds.³² Galabioside 13 was attached to the commercially available PAMAM dendrimer scaffold 36 containing eight end groups by use of the TBTU coupling reagent [O-(ben-

Table 1. Inhibition of the S. suis Mediated Hemagglutination

 of Human Erythrocytes by Galabiose Derivatives

	S. suis 628 (P_N)		S. suis 836 (P ₀)	
inhibitor	MIC [nM]	rel. pot. ^a (per sugar) ^b	MIC [nM]	rel. pot. ^a (per sugar) ^b
22 monovalent	900	1(1)	1600	1 (1)
25 divalent	6	150(75)	70	22(11)
32 divalent, long arms	9	100(50)	100	16 (8)
29 divalent rigid 1,4	7	128 (64)	90	18 (9)
27 divalent rigid 1,3	4	225(112)	50	32(16)
35 tetravalent	2	450 (112)	40	40 (10)
38 octavalent, PAMAM	0.3	3000 (375)	2	800 (100)

 a Relative potency = IC_{50}(monovalent **22**)/IC_{50}(multivalent compound). b Relative potency per sugar = relative potency/valency.

zotriazol-1-yloxy)-N,N,N',N'-tetramethyluronium tetrafluoroborate] and DIPEA in dimethyl sulfoxide (DMSO).³³ Compound **37** was purified on an LH20 column with CH₂Cl₂-MeOH (1:1) as the eluent. No free amino functions could be detected by the ninhydrin coloring reagent. Deacetylation by sodium methoxide in MeOH gave target compound **38** (89%) isolated after freeze-drying.

Inhibition Studies. The synthesized compounds were evaluated in two different assays. The first assay method was a hemagglutination of human erythrocytes, which express glycolipids with the Gala1-4Gal disaccharide on their surface.¹⁸ The S. suis bacteria induce agglutination that can be inhibited by galabiosides. The minimal inhibitory concentrations (MIC) required for complete inhibition of hemagglutination were determined for S. suis type P_N and P_O and are depicted in Table 1. MIC values in the low nanomolar range were obtained for both bacterial subtypes, indicating the great inhibitory activity of these compounds. For comparison of the inhibition strength of each of the multivalent compounds, the monovalent compound 22 was chosen as a reference compound for determining the multivalency effects. The monovalent compound had MIC values of 900 and 1600 nM for subtypes $P_{\rm N}$ and P₀, respectively. The MICs for the multivalent compounds were lower than for the monovalent reference compound, also when corrected for valency, indicating a multivalent or cluster effect.³⁴ Inhibition of the Po subtype required higher concentrations of inhibitors than for the P_N subtype, and also the response to multivalency effects was smaller for the P₀ subtype. Both of these effects may originate from the more strict carbohydrate specificity of its adhesin.¹⁸ The MIC of divalent compound 25 for the P_N subtype was 6 nM, that is, a 150-fold potency increase compared to reference compound 22. Elongation of the spacer arms (compound 32) and partial preorganization of the conformation of the divalent compounds by introducing a rigid core structure (compounds 27 and 29) was hoped to shed light on the favorable geometry of effective divalent binding. The compounds were, however, almost equally as effective as compound 25. By increasing the valency of the inhibitor molecule from two to four galabiose moieties (compound 35), a 3-fold increase in relative potency per galabioside was observed. However, the strongest multivalency effects were seen with the octavalent PAMAM compound 38. For this compound MIC with the P_N subtype of 0.3 nM was determined, which made it 3000-fold more potent (375-fold per sugar) than the reference compound 22. To the best of our





^a (a) BOP, DIPEA, CH₂Cl₂, DMF [however, TBTU, DIPEA, DMSO for **36**]; (b) NaOMe, MeOH.

knowledge, this is the first example of a soluble carbohydrate inhibiting bacterial binding at a subnanomolar concentration.

In the surface plasmon resonance (SPR) assay, pigeon ovomucoid, a glycoprotein rich in galabiose sequences, was immobilized on the sensor chip surface. Whole bacteria (S. suis 628, P_N) were added, and the SPR signal (in resonance units, RU) was detected. The whole bacteria showed a clear SPR signal, indicative of their association with the glycoprotein on the chip, and the signal could be inhibited in a dose-dependent manner by the synthesized galabiosides (Figure 1A). An inhibition curve was calculated for each compound and IC_{50} values were determined (Figure 1B; Table 2).

MIC values of the most potent compounds were again approaching the low nanomolar range, confirming their great inhibitory potency. The monovalent compound **22** proved to be a good inhibitor (IC₅₀ of 1800 nM), but increasing the valency to two galabiose moieties (**25**) led to an 8-fold more potent inhibitor per galabiose unit (Figure 1, Table 2). Again, increasing the length of spacer arms or partial preorganization of the conformation had no major effects in inhibition compared to compound **25**. By increasing the valency of the inhibitor from two to four galabiose moieties (compound **35**), a strong increase in inhibition potency was observed. An IC₅₀ of 7 nM was determined, which made it 256 times more potent than reference compound **22** and 33 times



Figure 1. Inhibition of whole bacteria (*S. suis* 628, P_N) binding to pigeon ovomucoid coated sensor chip in a surface plasmon resonance assay. (A) Binding of *S. suis* in the presence of various concentrations of monovalent galabiose compound (RU, resonance units). (B) Inhibition curves for monovalent **22** (\blacksquare), divalent **25** (\blacklozenge), and tetravalent **35** (×) galabiose compound (nRU, normalized resonance units).

Table 2. Inhibition of *S. suis* Binding to the Pigeon Ovomucoid Immobilized on a Surface Plasmon Resonance Chip by Galabiose Derivatives

	S. suis 628 (P_N)		
inhibitor	IC ₅₀ [nM]	rel. pot. ^{<i>a</i>} (per sugar) ^{<i>b</i>}	
22 monovalent	1800	1 (1)	
25 divalent	235	8 (4)	
32 divalent, long arms	156	12 (6)	
29 divalent rigid 1,4	309	6 (3)	
27 divalent rigid 1,3	293	6 (3)	
35 tetravalent	7	256 (64)	
38 octavalent, PAMAM	7	256 (32)	

^{*a*} Relative potency = IC_{50} (monovalent **22**)/ IC_{50} (multivalent compound). ^{*b*} Relative potency per sugar = relative potency/valency.

more potent than divalent compound **25**, indicating a strong multivalency effect (Figure 1, Table 2). The octavalent PAMAM compound **38** had inhibitory power at similar levels as compound **35**.

To further confirm the results of the SPR experiments, similar inhibition experiments were also performed by use of a previously reported enzyme-linked immunosorbent-type assay (ELISA) with pigeon ovomucoid coated plates (data not shown). The IC₅₀ values found in the ELISA assay were higher (IC₅₀ compound **22**, 45 μ M) than the MIC values of the hemagglutination and also higher than the IC₅₀ values of the SPR assay. It was also found that the differences between the compounds were smaller than in the previously mentioned assay. However, in agreement with the SPR assay, the tetravalent **35** was the most potent with a relative potency of 12.

Conclusions

A series of mono- and multivalent galabiosides were synthesized in high yields by coupling of one general galabiose building block to different scaffolds. The synthesized galabiosides were tested for their inhibition of the binding of the bacterial pathogen S. suis. Two different assays, hemagglutination and SPR, were used to determine the inhibitory concentrations of the compounds. For comparison of the inhibition strength of each multivalent compound, the monovalent compound 22 was used as a reference compound. In the hemagglutination assay, inhibition of two different subtypes of the adhesin of S. suis $(P_N \text{ and } P_O)$ were tested. All synthesized compounds proved to be excellent inhibitors of the binding of bacteria to galabiosylated surfaces. Strong multivalency effects could be clearly observed, with compounds of higher valency having lower MIC

and IC₅₀ values. In one case a MIC of 0.3 nM, for octavalent 38, was determined, representing an increase in potency of 3000-fold, or 375-fold per galabiose unit. This enhancement was much larger than the previously reported most effective compounds, a divalent and a tetravalant galabioside, which both showed a 25-fold enhancement on a per sugar basis for the same S. suis subtype and the same assay.¹⁰ Despite good agreement between the data from the two highly different assay methods, some differences were also observed. While the octavalent 38 was the most potent in the hemagglutination assay, the tetravalent 35 showed a higher relative potency per galabiose unit in the SPR assay. When the performance of the four divalent compounds was compared in the two assays, there was no strong correlation to spacer length, rigidity, and orientation (27) vs 29). Valency seems to be a stronger determining factor. This result has implications on the orientation and mobility of the different binding sites determining the multivalent enhancement. Our results may bring the anti-adhesion approach against bacterial infections closer to its potential, since strong inhibition was observed as caused by multivalent presentation of the binding galabiose moiety. Further improvements will be pursued by evaluating complementary orientations and valencies. Gratifyingly, we note that multivalent carbohydrates have successfully been used in vivo, as was recently shown in two separate studies focused on the neutralization of the Shiga toxin of E. $coli.^{35}$

Experimental Section

General Methods. All chemicals were of reagent grade and were used without further purification. Reactions were monitored by thin-layer chromatography (TLC) on silica gel 60 F₂₅₄ (Merck); after examination under UV light, compounds were visualized by heating with 10% (v/v) methanolic H_2SO_4 or with ninhydrin (3 mg mL^{-1}) . In the workup procedures of reaction mixtures, organic solutions were washed with appropriate amounts of the indicated aqueous solutions, then dried, and concentrated under reduced pressure at 40 °C. Column chromatography was performed on silica gel 60 (Merck, 0.063-0.200 mm). Final compounds were purified on an Isolute SPE column [IST, 291-1000-F C8(EC)]. ¹H NMR spectra were recorded at 300 K with a Varian Gemini-300 (300 MHz) or a Varian Unity INOVA 500 (500 MHz) spectrometer; the $\delta_{\rm H}$ values are given in parts per million (ppm) relative to the signal for internal Me_4Si ($\delta_{\rm H}$ = 0, CDCl_3 and CD_3OD) or internal acetone ($\delta_{\rm H} = 2.225$, D₂O). ¹³C NMR spectra (APT, 75 MHz) were recorded at 300 K with a Varian Gemini-300 spectrometer; $\delta_{\rm C}$ values are given in ppm relative to the signal of CDCl₃ ($\delta_{\rm C} = 76.9$, CDCl₃) or for internal acetone ($\delta_{\rm C} = 30.89$, D_2O) or internal MeOH ($\delta_C = 49.50$, D_2O). Two-dimensional

¹H-¹H correlation and total correlation spectroscopy (COSY and TOCSY) and ¹H-¹³C correlated heteronuclear single quantum coherence (HSQC) NMR spectra (500 MHz) were recorded at 300 K with a Varian Unity INOVA 500 spectrometer. Exact masses were measured by nanoelectrospray timeof-flight mass spectrometry on a Micromass LCToF mass spectrometer at a resolution of 5000 fwhm. Gold-coated capillaries were loaded with 1 μ L of sample (concentration 20 μ M) dissolved in a 1:1 (v/v) mixture of CH₃CN-H₂O with 0.1% formic acid. NaI or poly(ethylene glycol) (PEG) was added as internal standard. The capillary voltage was set between 1100 and 1350 V, and the cone voltage was set at 30 V. Matrixassisted laser desorption ionization time-of-flight (MALDI-TOF) spectra were recorded on a Shimadzu Axima-CFR with a-cyano-4-hydroxycinnamic acid (CHCA) as a matrix and adrenocorticotropin (ACTH) fragment 18-39 for calibration.

3-Azidopropyl 2,3,4,6-Tetra-O-acetyl-β-D-galactopyra**noside** (3). A solution of 1,2,3,4,6-penta-O-acetyl- β -D-galactopyranoside 1 (9.2 g, 23.7 mmol) and 3-bromopropanol (4.9 g, 35.5 mmol) in dry CH₂Cl₂ (100 mL), containing molecular sieve 4 Å (1.5 g), was cooled to 0 °C, and BF₃·Et₂O (15 mL, 118.5 mmol) was added in 2 min. The mixture was stirred overnight, neutralized with NEt₃, washed with aqueous saturated NaHCO3, H2O, and aqueous saturated NaCl, dried (Na2-SO₄), filtered, and concentrated. Reduced-pressure column chromatography (toluene-EtOAc 8:1) of the residue gave 2(9.36 g, 84%), isolated as a slightly yellow syrup. To a solution of 2 (0.81 g, 1.72 mmol) in dry DMF (5 mL) was added NaN₃ (0.56 g, 8.63 mmol), and the mixture was stirred at 100 °C overnight, then filtered over Hyflo, and coconcentrated with toluene. The residue was dissolved in dichloromethane and washed with aqueous saturated NaCl. dried (Na₂SO₄), filtered. and concentrated to give 3 (0.65 g, 87%), isolated as a slightly yellow syrup. ¹H NMR (300 MHz, CDCl₃): $\delta = 1.84 - 1.89$ (m, 2 H, OCH₂CH₂CH₂N₃), 1.98, 2.05, 2.07, and 2.15 (4 s, each 3 H, 4 COCH₃), 3.38 (t, 2 H, OCH₂CH₂CH₂N₃), 3.58–3.65 (m, 1 H, OCHHCH2CH2N3), 3.93-4.00 (m, 2 H, H-5 and OCHHCH2-CH₂N₃), 4.10–4.21 (m, 2 H, H-6a and H-6b), 4.50 (d, $J_{1,2} = 8.1$ Hz, 1 H, H-1), 5.03 (dd, $J_{2,3} = 10.5$ Hz, $J_{3,4} = 3.6$ Hz, 1 H, H-3), 5.18 (dd, 1 H, H-2), 5.39 (d, $J_{4,5} < 1$ Hz, 1 H, H-4); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 23.2$, 23.3 (2 C), and 23.4 (COCH₃), 31.6 (OCH₂CH₂CH₂N₃), 50.5 (OCH₂CH₂CH₂N₃), 63.9 (C-6), 69.1 (OCH₂CH₂CH₂N₃), 69.7, 71.5, 73.3, and 73.5 (C-2, C-3, C-4, and C-5), 103.9 (C-1), 172.1, 172.8, 172.9, and 173.1 (COCH_3); HRMS for $C_{17}H_{25}N_3O_{10}$ (M, 431.154) M + NH₄ found 449.189, calcd 449.1883.

3-Azidopropyl 4,6-Di-O-benzylidene-\beta-D-galactopyra**noside (5).** To a solution of **3** (2.6 g, 6.1 mmol) in MeOH (20 mL) and CH₂Cl₂ (6 mL) was added 30% methanolic NaOMe (0.5 mL). The mixture was stirred for 1 h, then neutralized with Dowex 50X8 (H⁺), filtered, and concentrated. Reducedpressure column chromatography (CH₂Cl₂-MeOH 10:1) of the residue gave 4 (1.56 g, 97%), isolated as a colorless syrup. High-resolution MS data for $C_9H_{17}N_3O_6$ (M, 263.1117): M + H found 264.121, calcd 264.1195. To a solution of 4 (4.2 g, 15.9 mmol) in dry DMF (20 mL) were added α, α -dimethoxytoluene (19 mL, 127.6 mmol) and a catalytic amount of *p*-TSOH. The mixture was stirred for 2.5 h at 50 °C and neutralized with NEt₃. After the addition of EtOAc, the mixture was washed with H₂O and aqueous saturated NaCl, dried (MgSO₄), filtered, and coconcentrated with toluene. Reduced-pressure column chromatography (toluene–EtOAc, $10:1 \rightarrow 1:4$) of the residue gave 5 (4.5 g, 80%), isolated as a white solid. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.91 - 1.95$ (m, 2 H, OCH₂CH₂CH₂N₃), 3.43-3.50 (m, 3 H, H-5 and OCH₂CH₂CH₂N₃), 3.62-3.79 (m, 3 H, H-2, H-3, and OCHHCH2CH2N3), 4.00-4.06 (m, 1 H, OCHH-CH₂CH₂N₃), 4.09 (dd, $J_{5,6b} = 2.1$ Hz, $J_{6a,6b} = 12.3$ Hz, 1 H, H-6b), 4.22 (d, $J_{3,4} = 3.3$, $J_{4,5} < 1$ Hz, 1 H, H-4), 4.30 (d, $J_{1,2} = 1.5$ Hz, 1 H, Hz, 1 7.5 Hz, 1 H, H-1), 4.34 (dd, $J_{5,6a} < 1$ Hz, 1 H, H-6a), 5.56 (s, 1 H, OCHC₆H₅), 7.36-7.38 and 7.49-7.52 (2 m, 5 H, OCHC₆H₅); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 30.8$ (OCH₂CH₂CH₂N₃), 50.1 $(OCH_2CH_2CH_2N_3)$, 68.4 and 70.9 (C-6 and $OCH_2CH_2CH_2N_3)$, 68.3, 73.1, 74.3, and 77.2 (C-2, C-3, C-4, and C-5), 103.0 and 104.7 (C-1 and OCHC₆H₅), 128.3, 130.0, 131.0, and 139.5 $(OCHC_{6}H_{5});\,HRMS$ for $C_{16}H_{21}N_{3}O_{6}\,(M,\,351.1430)\,M+H$ found 352.152, calcd 352.1508.

3-Azidopropyl 2,3-Di-O-benzyl-4,6-di-O-benzylidene- β -**D-galactopyranoside (6).** To a solution of **5** (2.1 g, 5.97 mmol) in dry DMF (40 mL) was added NaH (0.76 g, 19.1 mmol), and the mixture was stirred for 0.5 h. After the addition of TBAI (0.44 g, 1.19 mmol) and BnBr (1.8 mL, 15.5 mmol), the reaction was stirred for 5 h, and then MeOH was added and the mixture was coconcentrated with toluene. The residue was dissolved in EtOAc, washed with H₂O, aqueous saturated NaHCO₃, and H₂O, dried (Na₂SO₄), filtered, and concentrated. Reduced-pressure column chromatography (toluene-EtOAc, $40:1 \rightarrow 10:1$) of the residue gave **6** (2.3 g, 75%), isolated as a white solid. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.80-1.95$ (m, 2) H, OCH₂CH₂CH₂N₃), 3.41 (t, 2 H, OCH₂CH₂CH₂N₃), 3.55 (dd, $J_{2,3} = 9.6 \text{ Hz}, J_{3,4} = 3.6 \text{ Hz}, 1 \text{ H}, \text{H-3}), 3.58 - 3.64 \text{ (m, 2 H, H-5)}$ and OCHHCH₂CH₂N₃), 3.82 (dd, J_{1,2} = 8.1 Hz, 1 H, H-2), 3.97-4.02 (m, 2 H, H-6b and OCHHCH2CH2N3), 4.10 (d, 1 H, H-4), $4.27 \text{ (dd, } J_{5.6a} < 1 \text{ Hz}, J_{6a.6b} = 12.3 \text{ Hz}, 1 \text{ H}, \text{H-6a}), 4.36 \text{ (d, 1)}$ H, H-1), 4.75 and 4.83 (s and dd, each 2 H, 2 OCH₂C₆H₅), 5.48 (s, 1 H, $OCHC_6H_5$), 7.28–7.38 and 7.53–7.56 (2 m, 15 H, OCHC₆ H_5 and 2 OCH₂C₆ H_5); ¹³C NMR (75.5 MHz, CDCl₃) δ $= 31.0 (OCH_2CH_2CH_2N_3), 50.1 (OCH_2CH_2CH_2N_3), 68.2, 71.0,$ 73.7, and 77.1 (C-6, OCH₂CH₂CH₂N₃, and OCH₂C₆H₅), 68.2, 75.5, 80.2, and 81.0 (C-2, C-3, C-4, and C-5), 103.0 and 105.3 (C-1 and OCHC₆H₅), 128.3, 129.4-130.7, and 139.7-140.6 $(OCHC_6H_5)$; HRMS for $C_{30}H_{33}N_3O_6$ (M, 531.2369) M + Na found 554.2196, calcd 554.2267.

3-Azidopropyl 2,3,6-Tri-O-benzyl-*β*-D-galactopyranoside (7). To a solution of 6 (0.61 g, 1.15 mmol) and TES (1.1 mL, 6.9 mmol) in dry CH₂Cl₂ (10 mL) was added, in 1 min at 0 °C, TFA (0.5 mL, 6.9 mmol). The mixture was stirred for 5 h, during which the temperature was allowed to reach room temperature and then neutralized with aqueous saturated $NaHCO_3$. The organic layer was separated, dried (Na_2SO_4), filtered, and concentrated. Reduced pressure column chromatography (toluene-EtOAc 10:1) of the residue yielded 7 (0.47 g, 76%), isolated as a colorless syrup. ¹H NMR (300 MHz, $CDCl_3$) $\delta = 1.70 - 1.75$ (m, 2 H, $OCH_2CH_2CH_2N_3$), 3.22 (t, 2 H, $OCH_2CH_2CH_2N_3$), 3.75 (dd, $J_{2,3} = 9.3$ Hz, $J_{3,4} = 3.3$ Hz, 1 H, H-3), 3.86 (d, $J_{4,5}$ < 1 Hz, 1 H, H-4), 4.23 (d, $J_{1,2}$ = 7.5 Hz, 1 H, H-1), 4.44 and 4.55 (2 s, each 2 H, 2 $\text{OC}H_2\text{C}_6\text{H}_5)$, 4.68 (2 d, each 1 H, OCH₂C₆H₅), 7.16-7.22 (m, 15 H, 3 OCH₂C₆H₅); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 28.8$ (OCH₂CH₂CH₂N₃), 47.8 (OCH₂CH₂CH₂N₃), 66.0, 68.8, 71.8, 73.2, and 74.8 (C-6, OCH₂-CH₂CH₂N₃, and OCH₂C₆H₅), 66.3, 72.8, 78.5, and 80.2 (C-2, C-3, C-4, C-5), 103.2 (C-1), 127.2-128.0 and 137.4-138.1 $(OCHC_6H_5)$; HRMS for $C_{30}H_{35}N_3O_6$ (M, 533.2526) M + Na found 556.2400, calcd 556.2424.

3-Azidopropyl (2,3,4,6-Tetra-O-benzyl-α-D-galactopyranosyl)-2,3,6-tri-O-benzyl-β-D-galactopyranoside (9). To a solution of 7 (1.10 g, 2.05 mmol) and 2,3,4,6-tetra-O-benzyl- α -D-galactopyranoside trichloroacetimidate 8 (2.11 g, 3.08 mmol) in dry Et₂O (50 mL) was added, at 0 °C, TMSOTf (8 μ L, 0.46 mmol). The mixture was stirred for 1.5 h, during which the temperature was allowed to reach room temperature, then neutralized with NEt₃, and concentrated. Reducedpressure column chromatography (toluene-EtOAc 40:1) of the residue gave 9 (1.60 g, 73%), isolated as a slightly yellow syrup. ¹H NMR (300 MHz, CDCl₃) δ = 1.81–1.89 (m, 2 H, OCH₂CH₂- CH_2N_3), 3.38 (t, 2 H, OCH₂CH₂CH₂N₃), 3.66 (dd, $J_{1,2} = 7.8$ Hz, $J_{2,3} = 9.9$ Hz, 1 H, H-2), 4.30 (d, 1 H, H-1), 5.04 (d, $J_{1,2} = 2.1$ Hz, 1 H, H-1'), 7.14-7.37 (m, 35 H, 7 OCH₂C₆H₅); ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3) \delta = 28.9 (\text{OCH}_2\text{CH}_2\text{CH}_2\text{N}_3), 48.0 (\text{OCH}_2\text{-}$ $CH_2CH_2N_3$), 66.2, 67.6, 71.9, 72.0, 72.4, 72.7, 72.9, 73.4, 74.6, and 74.9 (C-6, C-6', OCH2CH2CH2N3, and OCH2C6H5), 69.0, 73.3, 74.3, 74.4, 76.3, 78.6, 78.7, and 80.6 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 100.2 and 103.5 (C-1 and C-1'), 127.0-128.1 and 137.7-138.6 (OCHC₆H₅); HRMS for C₆₄H₆₉-N₃O₁₁ (M, 1055.4932) M + Na found 1078.477, calcd 1078.491.

3-(*N*-tert-Butyloxycarbonyl)aminopropyl (2,3,4,6-Tetra-O-benzyl- α -D-galactopyranosyl)-2,3,6-tri-O-benzyl- β -D-galactopyranoside (10). A suspension of 9 (1.2 g, 1.13 mmol), Lindlar catalyst (0.24 g), Boc₂O (0.27 g, 1.25 mmol), and 3 drops of NEt3 in 1:1 MeOH-EtOAc (12 mL) was stirred overnight under an H₂ atmosphere. After the addition of EtOAc, the suspension was filtered over Hyflo and concentrated. Reduced-pressure column chromatography (toluene-EtOAc 15:1) of the residue yielded 10 (1.2 g, 92%), isolated as a colorless syrup. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.39$ [s, 9 H, OC(CH₃)₃], 1.78-1.82 (m, 2 H, OCH₂CH₂CH₂NH), 3.38 (dd, $J_{2,3} = 9.9$ Hz, $J_{3,4} = 2.7$ Hz, 1 H, H-3), 3.66 (dd, $J_{1,2} = 7.5$ Hz, 1 H, H-2), 4.00 (d, 1 H, H-4), 4.10 (d, $J_{1,2} = 2.1$ Hz, 1 H, H-2'), 4.30 (d, 1 H, H-1), 5.01 (d, 1 H, H-1'), 7.17-7.37 (m, 35 H, 7 $OCH_2C_6H_5$); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 28.2$ [OC(CH₃)₃], 29.6 (OCH₂CH₂CH₂NH), 37.5 (OCH₂CH₂CH₂NH), 67.4, 67.7, 71.9, 72.0, 72.7, 72.9, 73.4, 74.6, 74.9, and 78.6 (C-6, C-6', OCH₂CH₂CH₂NH, and OCH₂C₆H₅), 69.0, 73.3, 74.3, 74.4, 76.3, 78.5, 78.7, and 80.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 100.2 and 103.6 (C-1 and C-1'), 127.1-128.4 and 137.8-138.7 (OCHC6H5), 155.7 [OC(CH3)3]; HRMS for C69H79NO13 (M, 1129.5551) M + Na found 1152.542, calcd 1152.553.

3-(N-tert-Butyloxycarbonyl)aminopropyl (2,3,4,6-Tetra-O-acetyl- α -D-galactopyranosyl)-2,3,6-tri-O-acetyl- β -D-galactopyranoside (11). A solution of 10 (1.68 g, 1.48 mmol) and 3 drops of HOAc in 1:1 MeOH-EtOAc (12 mL) was stirred overnight under a H₂ atmosphere, then filtered over Hyflo, and concentrated. The residue was dissolved in 1:1 pyridine-acetic anhydride (100 mL) and stirred for 4 h, and then coconcentrated with toluene. Reduced-pressure column chromatography (toluene-EtOAc, $4:1 \rightarrow 1:1$) of the residue gave 11 (1.0 g, 88%), isolated as a white foam. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.43$ [s, 9 H, OC(CH₃)₃], 1.75-1.81 (m, 2 H, OCH₂CH₂CH₂NH), 1.98, 2.03, 2.06, 2.08, 2.10, and 2.13 (6 s, 3, 3, 3, 6, 3, and 3 H, 7 COCH₃), 3.26 (t, 2 H, OCH₂CH₂CH₂NH), 3.56 (m, 1 H, OCHHCH₂CH₂NH), 3.93 (m, 1 H, OCHHCH₂CH₂NH), 4.07 (d, $J_{3,4}=3.0~\mathrm{Hz},\,J_{4,5}<1~\mathrm{Hz},\,1~\mathrm{H},\,\mathrm{H}\text{-}4),\,4.57~\mathrm{(d},\,J_{1,2}=8.1~\mathrm{Hz},\,1$ H, H-1), 4.80 (dd, $J_{2,3} = 10.8$ Hz, 1 H, H-3), 5.00 (d, $J_{1,2} = 3.6$ Hz, 1 H, H-1'), 5.18 (dd, 1 H, H-2), 5.20 (dd, $J_{2,3} = 11.4$ Hz, 1 H, H-2'), 5.39 (dd, $J_{3,4} = 3.0$ Hz, 1 H, H-3'), 5.57 (d, $J_{4,5} < 1$ Hz, 1 H, H-4'); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 20.4-20.7$ $(COCH_3)$, 28.1 $[OC(CH_3)_3]$, 29.4 $(OCH_2CH_2CH_2NH)$, 37.4 (OCH_2-H_2NH) , 37.4 CH₂CH₂NH), 60.2 and 61.7 (C-6 and C-6'), 67.4 (OCH₂CH₂-CH₂NH), 66.7, 67.1, 67.5, 68.2, 68.3, 71.6, 72.4, and 76.8 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.1 and 100.8 (C-1 and C-1'), 155.7 [OC(CH₃)₃], 168.9, 169.6, 169.9, 170.1, 170.2, 170.3, and 170.4 (COCH₃); HRMS for C₃₄H₅₁NO₂₀ (M, 793.3004) M + Na found 816.286, calcd 816.298.

3-Aminopropyl (2,3,4,6-Tetra-O-acetyl-α-D-galactopyranosyl)-2,3,6-tri-O-acetyl-β-D-galactopyranoside Trifluoroacetic Acid Salt (12). To a solution of 11 (222 mg, 0.279 mmol) in CH₂Cl₂ (8 mL) was added, at 0 °C, 90% aqueous TFA (1 mL), and the mixture was stirred for 4 h and then coconcentrated with ethanol to give 12 (225 mg, quantitative), isolated as a white foam. ¹³C NMR (75.5 MHz, CDCl₃) δ = 20.4–20.7 (COCH₃), 28.1 [OC(CH₃)₃], 29.4 (OCH₂CH₂CH₂NH), 37.4 (OCH₂CH₂CH₂NH), 60.2 and 61.7 (C-6 and C-6'), 67.4 (OCH₂CH₂CH₂NH), 66.7, 67.1, 67.5, 68.2, 68.3, 71.6, 72.4, and 76.8 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.1 and 100.8 (C-1 and C-1'), 155.7 [OC(CH₃)₃], 168.9, 169.6, 169.9, 170.1, 170.2, 170.3, and 170.4 (COCH₃).

Short Spacer (2,3,4,6-Tetra-O-acetyl-α-D-galactopyranosyl)-2,3,6-tri-O-acetyl-β-D-galactopyranoside (13). To a solution of 12 (225 mg, 0.279 mmol) in pyridine (3 mL) and dry dioxane (1 mL) was added a solution of diglycolic anhydride (36 mg, 0.307 mmol) in dry dioxane (1 mL). The mixture was stirred overnight at 90 °C and then coconcentrated with toluene. The residue was dissolved in CH₂Cl₂ and washed with aqueous 1 M KHSO4 and aqueous saturated NaCl, dried (Na2- SO_4), filtered, and concentrated, yielding 13 (226 mg, quantitative), isolated as a white foam. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.82 - 1.89$ (m, 2 H, OCH₂CH₂CH₂NH), 2.00, 2.04, 2.06, 2.08, 2.09, 2.10, and 2.14 (7 s, each 3 H, 7 COCH₃), 3.81 (t, 2 H, OCH₂CH₂CH₂NH), 4.83 (dd, $J_{2,3} = 10.8$ Hz, $J_{3,4} = 2.7$ Hz, 1 H, H-3), 5.00 (d, $J_{1,2}$ = 3.6 Hz, 1 H, H-1'), 5.18 (dd, 1 H, H-2), 5.20 (dd, $J_{2,3} = 10.8$ Hz, 1 H, H-2'), 5.40 (dd, $J_{3,4} = 3.3$ Hz, 1 H, H-3'), 5.57 (d, $J_{4,5} < 1$ Hz, 1 H, H-4'); ¹³C NMR (75.5 MHz, $CDCl_3$) $\delta = 20.2 - 20.6 (COCH_3), 28.6 (OCH_2CH_2CH_2NH),$ 36.1 (OCH₂CH₂CH₂NH), 60.2 and 61.6 (C-6 and C-6'), 67.4 (OCH₂CH₂CH₂NH), 68.4 and 70.6 (COCH₂OCH₂COOH), 66.6, 67.1, 67.5, 68.1, 68.3, 71.5, 72.3, and 76.8 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.0 and 100.7 (C-1 and C-1'), 155.7 [OC(CH₃)₃], 169.2, 169.8, 169.9, 170.0, 170.1, 170.2, 170.3, 170.4, and 171.4 (COCH₃ and COCH₂OCH₂COOH); HRMS for C₃₃H₄₇NO₂₂ (M, 809.2590) M + Na found 832.2479, calcd 832.2488.

1.4-Bis[3-(tert-butoxycarbonyl)aminoprop-1-ynyl]ben**zene** (16). To a solution of 14 (2.0 g, 6.06 mmol), CuI (0.11 g, 0.6 mmol), and Pd⁽⁰⁾(PPh₃)₄ (0.34 g, 0.30 mmol) in dry CH₃CN (10 mL) was added, under N₂, a solution of Boc-protected propargylamine (2.8 g, 18.1 mmol) in dry NEt₃ (1.5 mL) and dry CH₃CN (1 mL). The reaction mixture was stirred under N₂ for 2 days, then diluted with CH₂Cl₂, filtrated over Hyflo, and concentrated. Reduced-pressure column chromatography (hexanes-EtOAc 15:1) of the residue gave 16 (1.77 g, 76%), isolated as a slightly brown solid. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.46$ [s, 18 H, 2 OC(CH₃)₃], 4.15 (d, J = 5.7 Hz, 4 H, 2 CH₂), 4.75 (br s, 2 H, 2 NH), 7.33 (s, 4 H, C₆H₄); ¹³C NMR $(75.5 \text{ MHz}, \text{CDCl}_3) \delta = 28.2 [OC(CH_3)_3], 31.8 (CH_2), 79.9$ [OC(CH₃)₃], 82.4 (CCH₂), 87.0 (CC₆H₄), 122.5 and 131.4 (C₆H₄), 155.1 (CO); HRMS for $C_{22}H_{28}N_2O_4$ (M, 384.2049) M + Na found 407.2028, calcd 407.1947.

Trifluoroacetic Acid Salt of 1,4-Bis(3-aminoprop-1-ynyl)benzene (17). To a solution of **16** (0.178 g, 0.463 mmol) in CH₂Cl₂ (10 mL) was added, at 0 °C, TFA (0.35 mL). The solution was stirred for 5 h, during which the temperature was allowed to reach room temperature, and then concentrated to give **17** (0.176 g, quantitative), isolated as a slightly brown solid. HRMS for $C_{12}H_{12}N_2$ (M, 184.1000) M + H found 185.1089, calcd 185.1078.

1,3-Bis[3-(tert-Butoxycarbonyl)aminoprop-1-ynyl]benzene (18). To a solution of 15 (1.0 g, 3.03 mmol), CuI (6 mg, 0.30 mmol), and Pd⁽⁰⁾(PPh₃)₄ (0.17 g, 0.151 mmol) in dry CH₃-CN (5 mL) was added, under N₂, a solution of Boc-protected propargylamine (1.41 g, 9.09 mmol) in dry NEt₃ (0.75 mL) and dry CH₃CN (1 mL). The reaction mixture was stirred under $N_{\rm 2}$ for 20 h, then diluted with $CH_{\rm 2}Cl_{\rm 2},$ filtrated over Hyflo, and concentrated. Reduced-pressure column chromatography (hexanes-EtOAc 15:1) of the residue gave 18 (1.02 g, 87%), isolated as a slightly yellow solid. ¹H NMR (300 MHz, CDCl₃) $\delta = 1.47$ [s, 18 H, 2 OC(CH₃)₃], 4.15 (d, J = 5.7 Hz, 4 H, 2 CH₂), 4.75 (br s, 2 H, 2 NH); ¹³C NMR (75.5 MHz, CDCl₃) $\delta =$ 28.2 [OC(CH₃)₃], 30.9 (CH₂), 79.9 [OC(CH₃)₃], 82.0 (CCH₂), 85.9 (CC₆H₄), 122.8, 128.2, 131.3, and 134.6 (C₆H₄), 155.1 (CO); HRMS for C₂₂H₂₈N₂O₄ (M, 384.2049) M + Na found 407.1951, calcd 407.1947.

Trifluoroacetic Acid Salt of 1,3-Bis(3-aminoprop-1-ynyl)benzene (19). To a solution of **18** (0.148 g, 0.385 mmol) in CH₂Cl₂ (8 mL) was added, at 0 °C, TFA (0.3 mL). The solution was stirred for 4 h, during which the temperature was allowed to reach room temperature, and then concentrated to give **19** (0.146 g, quantitative), isolated as a slightly yellow solid. ¹H NMR (300 MHz, CDCl₃) δ = 3.96 (s, 4 H, 2 CH₂), 4.55 (br s, 2 H, 2 NH), 7.32–7.54 (m, 4 H, C₆H₄); ¹³C NMR (75.5 MHz, CDCl₃) δ = 29.1 (CH₂), 80.2 (CCH₂), 85.2 (CC₆H₄), 121.4, 128.2, 131.8 and 134.4 (C₆H₄); HRMS for Cl₁₂H₁₂N₂ (M, 184.1000) M + Na found 185.1019, calcd 185.1078.

General Procedure for Preparation of Acetyl-Protected Galabiose Compounds 21, 24, 26, 28, 31, and 34. To a solution of 17, 19, 20, 23, 30, or 33, 13 (1.1 equiv per NH₂ group), and BOP (1.2 equiv per NH₂ group) in dry CH₂Cl₂ and dry DMF was added dropwise DIPEA (1.1 equiv per NH₂ group). The mixture was stirred and then concentrated. The residue was dissolved in EtOAc, washed with 1 M KHSO₄ and aqueous saturated NaCl, dried (Na₂SO₄), filtered, and concentrated. The residue was purified by reduced-pressure column chromatography (CH₂Cl₂-MeOH, $15:1 \rightarrow 1:1$) to give 21, 24, 26, 28, 31, or 34, isolated as white glass.

General Procedure for Preparation of Galabiose Compounds 22, 25, 27, 29, 32, and 35. To a solution of 21, 24, 26, 28, 31, or 34 in MeOH (1 mL) was added methanolic NaOMe (pH 9). The mixture was stirred for 1 h, then neutralized with Dowex 50X8 (H⁺), filtered, and concentrated. The residue was applied on a SPE-column, which was eluted with H₂O-AcCN (100:0 \rightarrow 85:15) to give **22**, **25**, **27**, **29**, **32**, or **35**, isolated as a white solid after freeze-drying.

Galabiose-Specific ¹H NMR data (300 MHz, CDCl₃) for Compounds 21, 24, 26, 28, 31, 34, and 37. δ = 4.48 (d, H-1), 4.83 (dd, H-3), 5.00 (d, H-1'), 5.17 (dd, 1H, H-2), 5.20 (dd, H-2'), 5.38 (dd, H-3'), 5.56 (d, H-4')

Monovalent Per-O-acetyl-Protected Galabiose Short Spacer (21). Reaction time 1 h; CH₂Cl₂/DMF 14:1; yield 66% (38 mg); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 20.4 - 20.7$ (COCH₃), 29.0 (OCH₂CH₂CH₂NH), 36.7 (OCH₂CH₂CH₂NH), 38.2 (OC-H₂CH₂NH), 52.0 (C₆H₄COCH₃), 60.3 and 61.6 (C-6 and C-6'), 66.5 and 68.0 (OCH₂CH₂CH₂NH and OCH₂CH₂NH), 70.9 and 71.0 (COCH₂OCH₂CO), 66.8, 67.1, 67.5, 68.3, 68.4, 71.8, 72.3, and 76.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.2 and 101.0 (C-1 and C-1'), 114.6, 119.5, 122.2, 129.4, 131.2, and 158.2 (C₆H₄COCH₃), 166.6 (C₆H₄COCH₃), 168.6-170.5 (COCH₃ and CONH); HRMS for C₄₃H₅₅N₂O₂₄ (M, 986.3380) M + Na found 1009.3435, calcd 1009.3278.

Monovalent Galabiose Short Spacer (22). Yield 80% (22 mg); ¹H NMR (300 MHz, D_2O) $\delta = 1.76$ (m, 2 H, OCH₂CH₂-CH₂NH), 3.27 (m, 2 H, OCH₂CH₂CH₂NH), 3.48 (dd, $J_{1,2} = 7.8$ Hz, $J_{2,3} = 10.2$ Hz, 1 H, H-2), 3.60–3.68 (m, 7 H, H-3, H-5, H-6a', H-6b', OCHHCH2CH2NH, and OCH2CH2NH), 3.73-3.79 (m, 3 H, H-2', H-6a, and H-6b), 3.83-3.87 (m, 5 H, H-3', OCHHCH₂CH₂NH, and C₆H₃COCH₃), 3.97 (s, 4 H, H-4, H-4', and COCH₂OCH₂CO), 4.03 (s, 2 H, COCH₂OCH₂CO), 4.17 (m, 2 H, OCH2CH2NH), 4.29-4.36 (m, 2 H, H-1 and H-5'), 4.90 $(d, J_{1,2} = 3.9 \text{ Hz}, 1 \text{ H}, \text{H-1'}), 7.17 (dd, 1 \text{ H}, \text{CH}^5), 7.38 (dd, 1 \text{ H}, \text{CH}^5)$ CH4), 7.46 (dd, 1 H, CH2), 7.56 (dd, 1 H, CH6), 8.16 and 8.37 (2 t, each 1 H, 2 NH); ¹³C NMR (75.5 MHz, D_2O) δ = 29.3 (OCH₂CH₂CH₂NH), 36.6 (OCH₂CH₂CH₂NH), 39.2 (OCH₂CH₂-NH), 53.3 (C₆H₄COCH₃), 60.7 and 61.1 (C-6 and C-6'), 67.2 (OCH₂CH₂NH), 68.2 (OCH₂CH₂CH₂NH), 69.4, 69.5, 69.7, 71.2, 71.4, 73.0, 75.6, 77.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 70.3 (2 C) (COCH₂OCH₂CO), 100.6 (C-1'), 103.5 (C-1), 115.6, 121.1, 122.9, and 130.5 (C₆H₄COCH₃), 168.7 (C₆H₄CO-CH₃), 172.0 and 172.4 (CONH); HRMS for C₂₉H₄₄N₂O₁₇ (M, 692.2640) M + Na found 715.2514, calcd 715.2538.

Divalent Per-O-acetyl-Protected Galabiose Short Spacer (24). Reaction time 1 h; CH_2Cl_2/DMF 8:1; yield 88% (135 mg); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 20.3-20.6$ (COCH₃), 28.9 (OCH₂CH₂CH₂NH), 36.5 (OCH₂CH₂CH₂NH), 38.1 (OC-H₂CH₂NH), 52.0 (C₆H₃COCH₃), 60.2 and 61.5 (C-6 and C-6'), 66.4 and 67.9 (OCH₂CH₂CH₂NH and OCH₂CH₂NH), 70.7 and 70.8 (COCH₂OCH₂CO), 66.7, 67.0, 67.4, 68.1, 68.3, 71.6, 72.2, and 76.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.0 and 100.8 (C-1 and C-1'), 106.3, 107.8, 133.8, and 159.2 (C₆H₃-COCH₃), 166.1 (C₆H₃COCH₃), 168.6-170.3 (COCH₃ and CONH); HRMS forC₇₈H₁₀₉N₄O₄₆ (M, 1836.6235) M + Na found 1859.687, calcd 1859.621.

Divalent Galabiose Short Spacer (25). Yield 99% (74 mg); ¹H NMR (300 MHz, D₂O) $\delta = 1.79 - 1.83$ (m, 4 H, 2 OCH₂CH₂CH₂NH), 3.30 (m, 4 H, 2 OCH₂CH₂CH₂NH), 3.35 (s, $3 \text{ H}, \text{ C}_6\text{H}_3\text{COCH}_3), 3.54 \text{ (dd}, J_{1,2} = 7.8 \text{ Hz}, J_{2,3} = 10.2 \text{ Hz}, 2 \text{ H},$ 2 H-2), 3.63-3.71 (m, 14 H, 2 each H-3, H-5, H-6a', H-6b', OCHHCH₂CH₂NH, and OCH₂CH₂NH), 3.78-3.93 (m, 10 H, H-2', H-3', H-6a, H-6b, and OCHHCH₂CH₂NH), 4.02 (d, J_{3,4} = 3.3 Hz, $J_{4,5}$ < 1 Hz, 2 H, 2 H-4), 4.03 (d, $J_{3,4}$ = 3.6 Hz, $J_{4,5}$ < 1 Hz, 2 H, 2 H-4'), 4.04 and 4.10 (2 s, each 4 H, 2 COCH₂-OCH₂CO), 4.13 (m, 4 H, 2 OCH₂CH₂NH), 4.36 (m, 2 H, 2 H-5'), 4.38 (d, 2 H, 2 H-1), 4.95 (d, $J_{1,2} = 3.6$ Hz, 2 H, 2 H-1'), 6.70 and 7.05 (2 s, 1 and 2 H, C₆H₃COCH₃); ¹³C NMR (75.5 MHz, D₂O): $\delta = 29.2$ (OCH₂CH₂CH₂NH), 36.6 (OCH₂CH₂CH₂NH), 39.1 (OCH₂CH₂NH), 53.4 (C₆H₃COCH₃), 60.7 and 61.1 (C-6 and C-6'), 67.2 (OCH₂CH₂NH), 68.2 (OCH₂CH₂CH₂NH), 69.3, 69.6, 69.8, 71.4, 71.6, 73.0, 75.7, 77.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 70.5 (2 C) (COCH₂OCH₂CO), 100.8 (C-1'), 103.5 (C-1), 107.5, 108.9, 132.0, and 159.9 (C₆H₃COCH₃), 168.7 (C₆H₃COCH₃), 172.0 (2 C) and 172.5 (2 C) (CONH); HRMS for $C_{50}H_{80}N_4O_{32}$ (M, 1248.4756) M + Na found 1271.4650, calcd 1271.4654.

Divalent Per-O-acetyl-Protected Short Spacer 1,3-Rigid Galabioside (26). Reaction time 2 h; CH₂Cl₂/DMF 7:1; yield 68% (101 mg); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 20.4-20.7$ (COCH₃), 28.9 (OCH₂CH₂CH₂H₂NH), 29.2 (CCH₂), 36.6 (OCH₂CH₂CH₂CH₂NH), 60.2 and 61.5 (C-6 and C-6'), 67.9 (OCH₂-CH₂CH₂NH), 70.8 (2 C, COCH₂OCH₂CO), 66.8, 67.1, 67.5, 68.2, 68.4, 71.7, 72.2, and 76.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 81.7 (CCH₂), 85.3 (CC₆H₄), 99.0 and 100.8 (C-1 and C-1'), 122.6, 128.1, 131.3, and 134.5 (CC₆H₄), 168.6-170.4 (COCH₃ and CONH); HRMS for C₇₈H₁₀₂N₄O₄₂ (M, 1766.5969) M + Na found 1789.6312, calcd 1789.5867.

Divalent Short Spacer 1,3-Rigid Galabioside (27). Yield 76% (51 mg); ¹H NMR (300 MHz, D2O) δ = 1.74 (m, 4 H, 2 OCH₂CH₂CH₂CH₂NH), 3.25 (m, 4 H, 2 OCH₂CH₂CH₂NH), 4.02 and 4.05 (2 s, each 4 H, 2 COCH₂OCH₂CO), 4.14 (s, 4 H, 2 CCH₂), 4.82 (d, $J_{1,2}$ = 2.7 Hz, 2 H, 2 H-1'), 7.28, 7.35, and 7.45 (3 m, 4 H, C₆H₄);¹³C NMR (75.5 MHz, D₂O) δ = 29.2 (OCH₂CH₂CH₂CH₂NH), 29.6 (CCH₂), 36.6 (OCH₂CH₂CH₂NH), 60.5 and 61.1 (C-6 and C-6'), 68.1 (OCH₂CH₂CH₂NH), 69.4, 69.6, 69.9, 71.4, 71.6, 73.0, 75.6, 77.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 70.5 (2 C, COCH₂OCH₂CO), 82.2 (CCH₂), 86.2 (CC₆H₄), 100.9 (C-1'), 103.5 (C-1), 172.1 (CONH); HRMS for C₅₀H₇₄N₄O₂₈ (M, 1178.4490) M + Na found 1201.4894, calcd 1201.4388.

Divalent Per-O-acetyl-Protected Short Spacer 1,4-Rigid Galabioside (28). Reaction time 1 h; CH_2Cl_2/DMF 5:1; yield 97% (87 mg); ¹³C NMR (75.5 MHz, $CDCl_3$) $\delta = 20.5-20.8$ (COCH₃), 28.9 (OCH₂CH₂CH₂NH), 29.2 (CCH₂), 36.9 (OCH₂CH₂CH₂NH), 60.3 and 61.6 (C-6 and C-6'), 68.1 (OCH₂-CH₂CH₂NH), 70.9 (2 C, COCH₂OCH₂CO), 66.9, 67.2, 67.5, 68.2, 68.5, 71.8, 72.3, and 76.8 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 82.3 (CCH₂), 86.3 (CC₆H₄), 99.1 and 100.9 (C-1 and C-1'), 168.8–170.4 (COCH₃ and CONH); HRMS for C₇₈H₁₀₂N₄O₄₂ (M, 1766.5969) M + Na found 1789.6218, calcd 1789.5867.

Divalent Short Spacer 1,4-Rigid Galabioside (29). Yield 93% (50 mg); ¹H NMR (300 MHz, D₂O) δ = 1.73 (m, 4 H, 2 OCH₂CH₂CH₂NH), 3.24 (m, 4 H, 2 OCH₂CH₂CH₂NH), 3.54 (dd, $J_{1,2}=7.8~{\rm Hz}, J_{2,3}=10.5~{\rm Hz}, 2~{\rm H}, 2~{\rm H-2}),\, 3.53-3.59~{\rm (m,~10~H},$ 2 each H-3, H-5, H-6a', H-6b', and OCHHCH₂CH₂NH), 3.66-3.79 (m, 10 H, 2 each H-2', H-3', H-6a, H-6b, and OCHHCH₂-CH₂NH), 3.78 (s, 4 H, 2 H-4 and 2 H-4'), 3.90 and 3.93 (2 s, each 4 H, 2 COCH₂OCH₂CO), 4.05 (s, 4 H, 2 CCH₂), 4.12-4.18 (m, 4 H, 2 H-1 and 2 H-5'), 4.82 (d, $J_{1,2} = 2.7$ Hz, 2 H, 2 H-1'), 7.26 (s, 4 H, C₆H₄); ¹³C NMR (75.5 MHz, D₂O) $\delta = 29.2$ $(OCH_2CH_2CH_2NH)$, 29.6 (CCH_2) , 36.6 $(OCH_2CH_2CH_2NH)$, 60.5 and 61.1 (C-6 and C-6'), 68.1 (OCH₂CH₂CH₂NH), 69.4, 69.6, 69.9, 71.4, 71.6, 73.0, 75.6, 77.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 70.5 (2 C, COCH2OCH2CO), 82.5 (CCH2), 87.1 (CC₆H₄), 101.0 (C-1'), 103.6 (C-1), 172.1 (CONH); HRMS for $C_{50}H_{74}N_4O_{28}$ (M, 1178.4490) M + Na found 1201.4495, calcd 1201.4388

Divalent Per-O-acetyl-Protected Galabiose Long Spacer (31). Reaction time 1 h; CH₂Cl₂/DMF 15:1; yield 61% (125 mg); ¹³C NMR (75.5 MHz, CDCl₃) $\delta = 20.4-20.6$ (COCH₃), 28.5 (2 C) and 28.9 (OCH₂CH₂CH₂NH), 36.3 and 36.8 (2 C, OCH₂-CH₂CH₂NH), 38.3 (OCH₂CH₂NH), 52.0 (C₆H₃COCH₃), 60.4 and 61.6 (C-6 and C-6'), 66.8, 67.1, 67.5, 68.1, 68.5, 71.7, 72.2, and 76.8 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 99.1 and 100.9 (C-1 and C-1'), 106.0, 107.8, 133.6, and 159.2 (*C*₆H₃-COCH₃), 166.3 (C₆H₃COCH₃), 169.4–170.3 (COCH₃ and CONH); HRMS for C₁₀₆H₁₆₀N₈O₅₈ (M, 2474.42, average mass) M + Na found 2495.6, calcd 2497.4.

Divalent Galabiose Long Spacer (32). Yield 77% (74 mg); ¹H NMR (300 MHz, D₂O) δ = 1.64–1.78 (m, 12 H, 6 OCH₂CH₂-CH₂NH), 3.18–3.26 (m, 12 H, 6 OCH₂CH₂CH₂NH), 4.28 (m, 2 H, 2 H-5'), 4.34 (d, J_{1,2} = 7.5 Hz, 2 H, 2 H-1), 4.95 (d, J_{1,2} = 3.0 Hz, 2 H, 2 H-1'), 6.62 and 6.99 (2 s, 1 and 2 H, C₆H₃COCH₃); ¹³C NMR (75.5 MHz, D₂O) δ = 28.8 (2 C), 29.1 (OCH₂CH₂CH₂NH), 36.6, 36.7 (2 C, 2 OCH₂CH₂CH₂NH), 39.1 (OCH₂CH₂CH₂NH), 53.4 (C₆H₃COCH₃), 60.6 and 61.0 (C-6 and C-6'), 67.2 (OCH₂CH₂NH), 68.2, 69.0 (2 C, 2 OCH₂CH₂CH₂CH₂CH₂ NH), 69.3, 69.5, 69.7, 71.3, 71.5, 72.9, 75.6, 77.6 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 69.9 and 70.1 (each 2 C, OCH₂ CH₂OCH₂CH₂O), 70.4 (2 C, COCH₂OCH₂CO), 100.8 (C-1'), 103.4 (C-1), 107.3, 108.8, 132.0, and 159.9 (C₆H₃COCH₃), 168.7 $(C_6H_3COCH_3),\,171.8,\,172.0,\,172.1,\,and\,172.4\,(CONH);\,HRMS$ for $C_{78}H_{132}N_8O_{44}$ (M, 1885.91, average mass) M + Na found 1908.9, calcd 1908.9

Tetravalent Per-O-acetyl-Protected Galabiose Short Spacer (34). Reaction time 4 h; CH₂Cl₂/DMF 15:1; yield 92% (224 mg); ¹³C NMR (75.5 MHz, CDCl₃) δ = 19.9–20.2 (COCH₃), 28.4 (OCH₂CH₂CH₂NH), 35.9 (OCH₂CH₂CH₂NH), 37.9 (OC-H₂CH₂NH), 53.0 (C₆H₃COCH₃), 60.3 and 61.6 (C-6 and C-6'), 65.6, 67.0, and 69.2 (OCH₂CH₂CH₂NH, OCH₂CH₂NH, OCH₂CH₂NH, and COCH₂OCH₂CO), 66.5, 67.0, 67.4, 67.9, 68.4, 71.5, 72.0, and 76.7 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 98.8 and 100.6 (C-1 and C-1'), 166.5 (C₆H₃COCH₃), 169.6–170.4 (COCH₃, CONH); HRMS for C₁₆₆H₂₂₆N₁₀O₉₄ (M, 3865.58, average mass) M + Na found 3888.6, calcd 3888.6.

Tetravalent Galabiose Short Spacer (35). Yield 79% (121 mg); ¹H NMR (300 MHz, D2O) δ = 1.73–1.88 (m, 8 H, 4 OCH₂CH₂CH₂CH₁NH), 3.23–3.37 (m, 8 H, 4 OCH₂CH₂CH₂CH₂NH), 4.30–4.38 (m, 8 H, 4 H-1 and 4 H-5'), 4.95 (s, 4 H, 4 H-1'), 6.64 and 6.77 (2 s, 3 and 6 H, 3 C₆H₃COCH₃); ¹³C NMR (75.5 MHz, D₂O) δ = 29.1 (OCH₂CH₂CH₂NH), 36.6 (OCH₂CH₂CH₂CH₂NH), 38.9 (OCH₂CH₂CH₂NH), 53.1 (C₆H₃COCH₃), 60.6 and 61.1 (C-6 and C-6'), 66.9 (OCH₂CH₂NH), 68.1 (OCH₂CH₂CH₂NH), 69.3, 69.5, 69.7, 71.3, 71.5, 73.0, 75.6, 77.6 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', and C-5'), 70.4 (COCH₂OCH₂OCH₂CO), 100.8 (C-1'), 103.5 (C-1), HRMS for C₁₁₀H₁₇₀N₁₀O₆₆ (M, 2688.55, average mass) M + Na found 2711.9, calcd 2711.6.

Octavalent Per-O-acetyl-Protected Short Spacer PAM-AM Generation 1 Galabioside (37). PAMAM generation 1 36 (20% weight solution in MeOH; 7.69 μ mol) was evaporated to dryness and dissolved in DMSO (5 mL). Compound 13 (54.8 mg, 67.7 μ mol), TBTU (23.7 mg, 73.8 μ mol), and DIPEA (62.4 μ L, 369 μ mol) were added and the mixture was stirred for 16 h and then concentrated. The residue was purified on a LH₂₀ column eluted with CH₂Cl₂–MeOH (1:1) to give 37 (34.7 mg, 58%), isolated as a white foam. MALDI-MS for C₃₂₆H₄₈₈N₃₄O₁₈₀ (M, 7763.48) M + Na found 7787.0, calcd 7786.5.

Bacterial Strains. The two galabiose-binding strains of *S.* suis, 628 and 836, have slightly different adhesin specificities as has been described previously.¹⁹ The bacteria were stored on THY-agar plates (Todd-Hewitt broth supplemented with 5% yeast extract) and cultivated in a CO₂ incubator. For binding assays they were cultivated in THY broth at 37 °C without shaking.

Hemagglutination. Hemagglutination assays were performed as described previously.³⁶ Briefly, equal volumes of bacteria and 5% sialidase-treated human erythrocytes were mixed and hemagglutination was visually recorded after 1-h incubation on ice. For inhibition assays, 2-fold dilutions of galabiose compounds $(25 \,\mu\text{L})$ were mixed with bacteria $(25 \,\mu\text{L})$. After 5 min of incubation at room temperature, 50 μ L of the erythrocytes was added. The hemagglutination was recorded as described and the MIC values (the lowest concentration completely inhibiting the hemagglutination) were recorded.

Surface Plasmon Resonance Studies. The SPR studies were performed with whole streptococcal cells.³⁷ Inhibition of *S. suis* 628 binding to pigeon ovomucoid, a glycoprotein rich in terminal galabiose structures, was measured with a BiaLite instrument (Biacore, Uppsala, Sweden). Pigeon ovomucoid was immobilized on an F1 sensor chip with Biacore amine coupling kit according to manufacturer's instructions. An empty channel, without any immobilized protein, was used as a reference.

The bacteria, suspended in HBS-P buffer [10 mM N-(2-hydroxyethyl)piperazine-N'-2-ethanesulfonic acid (HEPES), pH 7.4, 0.15 M NaCl, and 0.005% surfactant P20] to an OD₆₀₀ of 0.3, were injected over the sensor chip at a flow rate of 10 μ L/min and the binding to immobilized pigeon ovomucoid was recorded. For inhibition assays, dilutions of the galabiose compounds were made in HBS-P buffer and incubated with the bacteria for 5 min before injection over the sensor chip. Between the injections, the sensor chip surface was regenerated with 10 mM NaOH. For inhibition curve calculations the data from repeated experiments were normalized from 0 to 100 by use of the BIAevaluation program (version 3.0, Biacore). The IC₅₀ values, that is, the concentration of the compound that elicited half the maximal response, were defined for each compound.

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Supporting Information Available: NMR and mass spectra of the prepared compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- Levy, S. B. The challenge of antibiotic resistance. Sci. Am. 1998, 278 (March), 46–53.
 Johnson, K. F. Synthesis of oligosaccharides by bacterial en-
- Johnson, K. F. Synthesis of oligosaccharides by bacterial enzymes. *Glycoconjugate J.* **1999**, *16*, 141–146.
 Ofek, I.; Hasty, D. L.; Sharon, N. Anti-adhesion therapy of
- (3) Ofek, I.; Hasty, D. L.; Sharon, N. Anti-adhesion therapy of bacterial diseases: prospects and problems. *FEMS Immunol. Med. Microbiol.* 2003, *38*, 181–191.
- (4) Zopf, D.; Roth, S. Oligosaccharide anti-infective agents. Lancet 1996, 347, 1017–1021.
- (5) Mammen, M.; Choi, S.-K.; Whitesides, G. M. Polyvalent interactions in biological systems: Implications for design and use of multivalent ligands and inhibitors. *Angew. Chem., Int. Ed.* 1998, 37, 2754-2794.
- (6) (a) Kitov, P. I.; Sadowska, J. M.; Mulvey, G.; Arnstrong, G. D.; Ling, H.; Pannu, N. S.; Read, R. J.; Bundle, D. R. Shiga-like toxins are neutralized by tailored multivalent carbohydrate ligands. Nature 2000, 403, 669-672. (b) Kitov, P. I.; Shimizu, H.; Homans, S. W.; Bundle, D. R. Optimization of tether length in nonglycosidically linked bivalent ligands that target sites 2 and 1 of a Shiga-like toxin. J. Am. Chem. Soc. 2003, 125, 3284-3294. (c) Fan, E.; Zhang, Z.; Minke, W. E.; Hou, Z.; Verlinde, C. L. M. J.; Hol, W. G. J. High-affinity pentavalent ligands of Escherichia coli heat-labile enterotoxin by modular structurebased design. J. Am. Chem. Soc. 2000, 122, 2663-2664. (d) Zhang, Z.; Merritt, E. A.; Ahn, M.; Roach, C.; Hou, Z.; Verlinde, C. L. M. J.; Hol, W. G. J.; Fan, E. Solution and crystallographic studies of branched multivalent ligands that inhibit the receptorbinding of cholera toxin. J. Am. Chem. Soc. 2002, 124, 12991-12998. (e) Arosio, D.; Vrasidas, I.; Valentini, P.; Liskamp, R. M. J.; Pieters, R. J.; Bernardi, A. Synthesis and cholera toxin binding properties of multivalent GM1 mimics. Org. Biomol. Chem. 2004, 2, 2113-2124.
- Chem. 2004, 2, 2113-2124.
 (7) (a) Lindhorst, T. K.; Kieburg, C.; Krallmann-Wenzel, U. Inhibition of the type 1 fimbriae-mediated adhesion of *Escherichia coli* to erythrocytes by multiantennary alpha-mannosyl clusters: The effect of multivalency. *Glycoconjugate J.* 1998, *15*, 605-613. (b) Lindhorst, T. K.; Dubber, M.; Krallmann-Wenzel, U.; Ehlers, S. Cluster mannosides as inhibitors of type 1 fimbriae-mediated adhesion of *Escherichia coli*: Pentaerythritol derivatives as scaffolds. *Eur. J. Org. Chem.* 2000, 2027-2035. (c) Lindhorst, T. K.; Kötter, S.; Krallmann-Wenzel, U.; Ehlers, S. Trivalent alpha-D-mannoside clusters as inhibitors of type-1 fimbriae-mediated adhesion of *Escherichia coli*: structural variation and biotinylation. *J. Chem. Soc., Perkin Trans 1* 2001, 823-831.

- (8) Nagahori, N.; Lee, R. T.; Nishimura, S.-I.; Pagé, D.; Roy, R.; Lee, Y. C. Inhibition of adhesion of type 1 fimbriated *Escherichia coli* to highly mannosylated ligands. *ChemBioChem* **2002**, *3*, 836– 844.
- (9) Autar, R.; Khan, A. S.; Schad, M.; Hacker, J.; Liskamp, R. M. J.; Pieters, R. J. Adhesion inhibition of F1C-fimbriated *Escheri-chia coli* and *Pseudomonas aeruginosa* PAK and PAO by multivalent carbohydrate ligands. ChemBioChem 2003, 4, 1317-1325
- (10) Hansen, H. C.; Haataja, S.; Finne, J.; Magnusson, G. Di-, tri-, and tetravalent dendritic galabiosides that inhibit hemagglutination by Streptococcus suis at nanomolar concentration. J. Am. Chem. Soc. 1997, 119, 6974-6979.
- (11) Arends, J. P.; Zanen, H. C. Meningitis caused by Streptococcus suis in humans. Rev. Infect. Dis. 1988, 10, 131–137.
- Staats, J. J.; Feder, I.; Okwumabua, O.; Chengappa, M. M. Streptococcus suis: Past and present. Vet. Res. Commun. 1997, (12)21, 381-407
- (13) (a) Bock, K.; Breimer, M. E.; Brignoleg, A.; Hansson, G. C.; Karlssong, K.-A.; Larson, G.; Leffler, H.; Samuelsson, B. E.; Stromberg, N.; Eden, C. S.; Thurin, J. Specificity of binding of a strain of uropathogenic Escherichia coli to Gal alpha 1-4Galcontaining glycosphingolipids. J. Biol. Chem. 1985, 260, 8545-8551. (b) Roberts, J. A.; Marklund, B. I.; Ilver, D.; Haslam, D.; Kaack, M. B.; Baskin, G.; Louis, M.; Mollby, R.; Winberg, J.; Normark, S. The Gal(alpha-1-4)Gal-specific tip adhesin of Escherichia coli P-fimbriae is needed for pyelonephritis to occur in the normal urinary tract. Proc. Natl. Acad. Sci. U.S.A. 1994, 91, 11889-11893. (c) Stapleton, A. E.; Stroud, M. R.; Hakomori, S. I.; Stamm, W. E. The globoseries glycosphingolipid sialosyl galactosyl globoside is found in urinary tract tissues and is a preferred binding receptor in vitro for uropathogenic Escherichia coli expressing pap-encoded adhesins. Infect. Immun. 1998, 66, 3856 - 3861
- (14) (a) Gilboa-Garber, N.; Sudakevitz, D.; Sheffi, M.; Sela, R.; Levene, C. Paid interactions with the APO(H)-blood and P-blood group glycosphingolipid antigens may contribute to the broadspectrum adherance of Pseudomonas aeruginosa to human tissues in secondary infections. Glycoconjugate J. 1994, 11, 414-417. (b) Chen, C.-P.; Song, S.-C.; Gilboa-Garber, N.; Chang, K. S. S.; Wu, A. M. Studies on the binding site of the galactose-specific agglutinin PA-IL from *Pseudomonas aeruginosa*. *Gly*cobiology 1998, 8, 7-16.
- (15) Lindberg, A. A.; Brown, J. E.; Stromberg, N.; Westerling-Ryd, M.; Schultz, J. E.; Karlsson, K.-A. Identification of the carbo-hydrate receptor for Shiga toxin produced by *Shigella dysenteriae* type 1. J. Biol. Chem. 1987, 262, 1779–1785.
 (16) Lingwood, C. A.; Law, H.; Richardson, S.; Petric, M.; DeGrandis,
- S.; Brunton, J. L.; Karmali, M. Glycolipid binding of purified and recombinant *Escherichia coli* produced verotoxin in vitro. J. Biol. Chem. 1987, 262, 8834-8839.
- (17) Haataja, S.; Tikkanen, K.; Liukkonen, J.; François-Gerard, C.; Finne, J. Characterization of a novel bacterial adhesion specificity of Streptococcus suis recognizing blood group P receptor oligosaccharides. J. Biol. Chem. **1993**, 268, 4311–4317. (18) Haataja, S.; Tikkanen, K.; Nilsson, U.; Magnusson, G.; Karlsson,
- K.-A.; Finne, J. Oligosaccharide-receptor interaction of the Gala1-4Gal binding adhesin of Streptococcus suis combining site architecture and characterization of 2 variant adhesin specificities. J. Biol. Chem. 1994, 269, 27466-27472.
- (19) Haataja, S.; Zhang, Z.; Tikkanen, K.; Magnusson, G.; Finne, J. Determination of the cell adhesion specificity of Streptococcus suis with the complete set of monodeoxy analogues of globotriose. *Glycoconjugate J.* **1999**, *16*, 67–71.
- (20) Fan, E.; Merritt, E. A.; Verlinde, C. L. M. J.; Hol, W. G. J. AB₅ toxins: structures and inhibitor design. Curr. Opin. Struct. Biol. 2000, 10, 680-686.
- (21) Motawia, M. S.; Olsen, C. E.; Denyer, K.; Smith, A. M.; Moller, B. L. Synthesis of 4'-O-acetyl-maltose and α-D-galactopyranosyl-(1-4)-D-glucopyranose for biochemical studies of amylose biosynthesis. Carbohydr. Res. 2001, 330, 309-318.
- (22) Gandolfi-Donadio, L.; Gallo-Rodriguez, C.; de Lederkremer, R. M. Synthesis of α-D-Ga1p-(1-3)-β-D-Ga1f-(1-3)-D-Man, a terminal trisaccharide of Leishmania type-2 glycoinositolphospholipids. J. Org. Chem. **2002**, 67, 4430–4435. (23) De Ninno, M. P.; Etienne, J. B.; Duplantier, K. C. A method for
- the selective reduction of carbohydrate 4,6-O-benzylidene acetals. Tetrahedron Lett. 1995, 36, 669-672.

- (24) Wulff, G.; Röhle, G. Results and problems of O-glycoside (21) Wull, C., Hown, G. Desatts and probabilis of Optimistic synthesis. Angew. Chem., Int. Ed. Engl. 1974, 13, 157–170.
 (25) Reddy, P. G.; Pratap, T. V.; Kumar, G. D. K.; Mohanty, S. K.;
- Baskaran, S. The Lindlar catalyst revitalized: A highly chemoselective method for the direct conversion of azides to N-(tert-butoxycarbonyl)amines. Eur. J. Org. Chem. 2002, 3740-3743.
 (26) Mulders, S. J. E.; Brouwer, A. J.; Liskamp, R. M. J. Molecular
- diversity of novel amino acid based dendrimers. Tetrahedron Lett. 1997, 38, 3085-3088.
- (27) Brouwer, A. J.; Mulders, S. J. E.; Liskamp, R. M. J. Convergent synthesis and diversity of amino acid based dendrimers. Eur. J. Org. Chem. 2001, 1903–1915.
- (a) Vrasidas, I.; de Mol, N. J.; Liskamp, R. M. J.; Pieters, R. J. (28)Synthesis of lactose dendrimers and multivalency effects in binding to the cholera toxin B subunit. Eur. J. Org. Chem. 2001, 4685-4692. (b) André, S.; Pieters, R. J.; Vrasidas, I.; Kaltner, H.; Kuwabara, I.; Liu, F.-T.; Liskamp, R. M. J.; Gabius, H.-J. Wedgelike glycodendrimers as inhibitors of binding of mammalian galectins to glycoproteins, lactose maxiclusters, and cell surface glycoconjugates. ChemBioChem 2001, 2, 822-830. (c) Pieters, R. J. Interference with Lectin Binding and Bacterial Adhesion by Multivalent Carbohydrates and Peptidic Carbohydrate Mimics. Trends Glycosci. Glycotechnol. 2004, 16, 243–254.
- (29) Marshall, J. A.; Zou, D. A new synthesis of 3-carboxy-2,5disubstituted furans and their conversion to 5-vinyl derivatives. Tetrahedron Lett. 2002, 41, 1347–1350. Vrasidas, I.; Kemmink, J.; Liskamp, R. M. J.; Pieters, R. J.
- (30)Synthesis and cholera toxin binding properties of a lactose-2-aminothiazoline conjugate. Org. Lett. **2002**, *4*, 1807–1808.
- Tomalia, D. A.; Naylor, A. M.; Goddard, W. A., III Starburst dendrimers Molecular level control of size, shape, surface (31)chemistry, topology, and flexibility from atoms to macroscopic matter. Angew. Chem., Int. Ed. Engl. **1990**, 29, 138–175. (32) (a) Aoi, K.; Tsutsumiuchi, K.; Yamamoto, A.; Okada, M. Globular
- carbohydrate macromolecule "sugar balls" 3. "Radial-growth polymerization" of sugar-substituted α-amino acid N-carboxyanhydrides (GlycoNCAs) with a dendritic initiator. Tetrahedron 1997, 53, 15415-15427. (b) Page, D.; Roy, R. Synthesis and biological properties of mannosylated starburst poly(amidoamine) dendrimers. Bioconjugate Chem. 1997, 8, 714-723. (c) Zanini, D.; Roy, R. Practical synthesis of starburst PAMAM α-thiosialodendrimers for probing multivalent carbohydrate-lectin binding properties. J. Org. Chem. 1998, 63, 3486-3491.
 (e) Choudhury, A. K.; Kitaoka, M.; Hayashi, K. Synthesis of a cellobiosylated dimer and trimer and of cellobiose-coated polyamidoamine (PAMAM) dendrimers to study accessibility of an enzyme, cellodextrin phosphorylase. Eur. J. Org. Chem. 2003, 2462-2470. (f) Woller, E. K.; Walter, E. D.; Morgan, J. R.; Singel, D. J.; Cloninger, M. J. Altering the strength of lectin binding interactions and controlling the amount of lectin clustering using mannose/hydroxyl-functionalized dendrimers. J. Am. Chem. Soc. 2003, 125, 8820-8826.
- (33) Baek, M.-G.; Roy, R. Synthesis and protein binding properties of T-antigen containing GlycoPAMAM dendrimers. Bioorg. Med. Chem. 2002, 10, 11–17. Lundquist, J. J.; Toone, E. J. The cluster glycoside effect. Chem.
- Rev. 2002, 102, 555-578.
- (a) Mulvey, G. L.; Marcato, P.; Kitov, P. I.; Sadowska, J.; Bundle, (35)D. R.; Armstrong, G. D. Assessment in mice of the therapeutic potential of tailored, multivalent Shiga toxin carbohydrate ligands. J. Infect. Dis. 2003, 187, 640-649. (b) Nishikawa, K.; Matsuoka, K.; Kita, E.; Okabe, N.; Mizuguchi, M.; Hino, K.; Miyazawa, S.; Yamasaki, C.; Aoki, J.; Takashima, S.; Yamakawa, Y.; Nishijima, M.; Terunuma, D.; Kuzuhara, H.; Natori, Y. A therapeutic agent with oriented carbohydrates for treatment of infections by Shiga toxin-producing Escherichia coli O157: H7. Proc. Natl. Acad. Sci. U.S.A. 2002, 99, 7669–7674.
- (36) Korhonen, T. K; Finne, J. In Enterobacterial surface antigens: methods for molecular characterization; Korhonen, T. K., Dawes, E. A., Mäkelä, P. H., Eds.; Elsevier Science Publishers: Am-sterdam, 1985; pp 301–313.
- Senpuku, H.; Matin, K.; Abdus, S. M. D.; Kurauchi, I.; Sakurai, S.; Kawashima, M.; Murata, T.; Miyazaki, H.; Hanada, N. (37)Inhibitory effects of MoAbs against a surface protein antigen in real-time adherence in vitro and recolonization in vivo of Streptococcus mutans. Scand. J. Immunol. 2001, 54, 109-116.

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